

Humidity

- Humidity is defined simply as the amount of water vapor in the air.
- Relative humidity is the most familiar measure of water vapor content.
 - The problem with relative humidity and the reason it is not used in meteorological analysis is that it is dependent on temperature.
 - Even if the moisture content is unchanged throughout the day, the relative humidity will change because of its dependence on temperature.
- A moisture index which is not dependent on temperature is known as the 'dew point' (T_D).
 - The temperature at which the atmosphere becomes saturated is the dew point temperature.
 - The higher the dew point temperature, the more water vapor that is in the air.
 - If the air temperature is equal to the dew point temperature, the relative humidity is equal to 100%.
 - The dew point temperature can *never* exceed the air temperature.
 - The dew point temperature is usually given in degrees Fahrenheit.
- Assuming the air temperature is 80°F or above.
 - If the dew point temperature is:
 - 80°F or higher ---→oppressive
 - 70°F - 79°F ----→very uncomfortable
 - 60°F - 69°F ----→uncomfortable
 - 50°F - 59°F ---→comfortable
 - Lower than 50°F -----→very comfortable

Adiabatic Processes

- An 'adiabatic process' is one in which temperature changes but no heat is added to or removed from a substance.
- Important to the formation of clouds.
- In an adiabatic process, expanding air cools and air being compressed warms.
- As a parcel rises, it expands due to lower atmospheric pressure surrounding it.
 - Air pressure decreases the further you get from Earth's surface.
- As a parcel descends toward Earth, it compresses due to higher atmospheric pressure surrounding it.
 - Air pressure increases the closer you get to Earth's surface.

- When a parcel expands, there are fewer molecular collisions within the parcel because of the greater space the parcel occupies. As a result, temperature decreases.
- When a parcel compresses, there are more molecular collisions within the parcel because of the lesser space the parcel occupies. As a result, temperature increases.
- The rate at which a rising air parcel cools is known as the ‘adiabatic lapse rate’ and is equal to $1.0^{\circ}\text{C}/100\text{ m}$ (or $5.5^{\circ}\text{F}/1000\text{ ft.}$).

Environmental Lapse Rate

- The ‘environmental lapse rate’ (ELR) is different from the ‘adiabatic lapse rate’ in that the environmental lapse rate measures temperature changes in the vertical for a large mass of air, instead of an air parcel.
 - In other words the environmental lapse rate measures the overall decrease in temperature with increasing height.
 - The value of the ‘ELR’ in the troposphere is variable and does not have a fixed value.

Forms of Condensation

- Dew is a liquid which condenses on a surface.
 - Most likely to form on clear, windless nights.
 - When the temperature cools to the dew point, dew forms.
- Frost is similar to dew except that it occurs when the temperature is below freezing.
 - Frost involves the phase change from water vapor to ice, without going through the liquid phase.
- Fog is a cloud at Earth’s surface.
 - Forms when the air temperature lowers to the dew point.
- Steam fogs occur when very cold, dry air interacts with warm, moist air over a body of water.
- Radiation fogs occur when the temperature at night cools to the dew point.
 - Most likely to form on cloudless nights.
 - Dissipate within a few hours of sunrise.
 - Forms often in the Central Valley of California.
- Advection fogs develop when warm, moist air moves horizontally (i.e., advects) over a cooler surface.
 - The air as a result cools.
 - If the air cools to the dew point, fog forms.
 - Common during the summer months over the San Francisco Bay area.
- Upslope fog develops when air moves upward along a mountain range.
 - As a result, it expands and cools adiabatically.
 - If the air cools to the dew point, fog forms.

- Common along the eastern edge of the Rocky Mountains in Colorado.
- The city of Denver, Colorado, being on the foothills of the Rocky Mountains, frequently observes upslope fogs.

An Aside

- As an air parcel expands, it cools due to fewer molecular collisions. BUT:
 - Why is cold air denser than warm air considering that as volume increases, density decreases as prescribed by 'Density = Mass/Volume' ($D = M/V$)?
- It has to relate to the speed of the molecules.
- Warm air molecules move at a higher speed than cold air molecules. This is due to the kinetic energy of the molecules.
- Warm air molecules take up a higher spatial magnitude than cold air molecules.
- As a result, the density of warm air is lower than that of cold air.

Helpful Links:

[http://ww2010.atmos.uiuc.edu/\(Gh\)/guides/maps/sfcobs/dwp.rxml](http://ww2010.atmos.uiuc.edu/(Gh)/guides/maps/sfcobs/dwp.rxml)

http://apollo.lsc.vsc.edu/classes/met130/notes/chapter6/adiab_cool.html

http://apollo.lsc.vsc.edu/classes/met130/notes/chapter6/adiab_warm.html